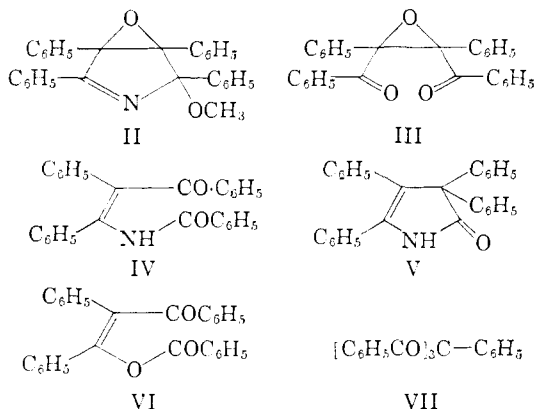


FORMATION OF EPOXIDES IN THE PYRROLE AND FURAN SERIES BY PHOTOOXIDATION

Sir:

Although the oxidation of indoles by air and light has been studied extensively,<sup>1,2,3</sup> little is known about the course of pyrrole photooxidation.<sup>4,5,6,7</sup> 2,3,4,5-Tetraphenylpyrrole (I), for example, has been reported<sup>7</sup> to yield only tars when irradiated with oxygen. We now have found that under *very mild conditions, i.e.*, exposure to air in methanolic solution during irradiation with a 150-watt reflector flood light, in the presence of methylene blue, 2,3,4,5-tetraphenylpyrrole undergoes ready oxidation, yielding products in 85% yield.

The main product (55%) is the epoxide, 5-methoxy-3,4-epoxy-2,3,4,5-tetraphenyl- $\Delta^1$ -pyrroline (II), m.p. 164–165° [Anal. Calcd. for C<sub>29</sub>H<sub>28</sub>NO<sub>2</sub>: C, 83.43; H, 5.55; N, 3.36; OCH<sub>3</sub>, 7.43. Found: C, 83.45; H, 5.43; N, 3.15; OCH<sub>3</sub>, 7.42]. II shows no OH, NH or carbonyl absorption in the infrared, is reduced to I with zinc in acetic acid, and is converted by 1 N HCl to *cis*-dibenzoylstilbene oxide (III), m.p. 172–173° [Anal. Calcd. for C<sub>28</sub>H<sub>20</sub>O<sub>3</sub>: C, 83.15; H, 4.98. Found: C, 83.25; H, 5.09]. Proof of structure of III, hitherto



unreported, rests on its infrared and ultraviolet absorption spectra<sup>8</sup> ( $\lambda_{\text{max}}^{\text{CHCl}_3}$  5.93  $\mu$ ;  $\lambda_{\text{max}}^{\text{Et}_2\text{O}}$  249

(1) P. M. Ray, *Ann. Rev. Plant Physiol.*, **9**, 81 (1958).

(2) B. Witkop, *et al.*, *Experientia*, **8**, 36 (1952), and references cited therein.

(3) Z. Yoshida and M. Kato, *THIS JOURNAL*, **76**, 311 (1954), and references cited therein.

(4) W. Metzger and H. Fischer, *Ann.*, **527**, 1 (1937).

(5) F. Bernheim and J. Morgan, *Nature*, **144**, 290 (1939).

(6) R. H. Linnel and S. Umar, *Arch. Biochem. and Biophys.*, **57**, 264 (1955).

(7) J. Martel, *Compt. rend.*, **244**, 626 (1957).

$m\mu$ ,  $\epsilon$  30,000) conversion to tetraphenylfuran with hydrogen iodide in glacial acetic acid, and reduction to dibenzoyldiphenylethane in ethanolic sodium hydrosulfite.<sup>9</sup>

The second oxidation product from I (30%) was identified as  $\alpha$ -N-benzoylamino- $\alpha'$ -benzoylstilbene (IV), m.p. 191° [Anal. Calcd. for C<sub>28</sub>H<sub>21</sub>NO<sub>2</sub>: C, 83.35; H, 5.25; N, 3.47. Found: C, 83.18; H, 5.43; N, 3.47]. IV shows infrared bands at 5.95  $\mu$  and 6.04  $\mu$ , can be hydrolyzed to dibenzoylphenylmethane, and was synthesized independently by benzylation of  $\alpha$ -amino- $\alpha'$ -benzoylstilbene.<sup>10</sup>

Photooxidation in the presence of potassium hydroxide led to a rearrangement of the benzoic acid type<sup>11</sup> with formation of the lactam, V (35%), m.p. 213–215° ( $\lambda_{\text{max}}^{\text{CHCl}_3}$  2.95  $\mu$ , 3.19  $\mu$  and 5.85  $\mu$ ) identical with the product<sup>12</sup> independently prepared from 2,2,3,4-tetraphenylcrotonolactone and ammonia.

Epoxide formation also was observed in the photooxidation of tetraphenylfuran. Whereas the oxidation in methanol led chiefly to *cis*-dibenzoylstilbene and 2,5-dimethoxy-2,3,4,5-tetraphenyl-2,5-dihydrofuran, both III (45%) and VI (20%), m.p. 151–152°, were formed during photooxidation in acetone. Compound VI ( $\lambda_{\text{max}}^{\text{CHCl}_3}$  5.75  $\mu$ , 5.99  $\mu$ ) is identical with the enol benzoate recently shown<sup>13</sup> to result from the benzylation of dibenzoylphenylmethane.<sup>14</sup>

The details of this work and discussion of the mechanisms involved will be reported separately.

CONTRIBUTION NO. 1602 FROM  
THE STERLING CHEMISTRY LABORATORY  
YALE UNIVERSITY  
NEW HAVEN, CONN.

HARRY H. WASSERMAN  
ARNO LIBERLES

RECEIVED MARCH 7, 1960

(8) H. House and D. J. Reif report  $\lambda_{\text{max}}$  252  $m\mu$ ,  $\epsilon$  15,100 for  $\alpha$ -phenyl-*cis*-benzalacetophenone oxide, *THIS JOURNAL*, **77**, 6525 (1955).

(9) R. E. Lutz and F. W. Wilder, *ibid.*, **56**, 2065 (1934).

(10) E. P. Kohler and N. K. Richtmyer, *ibid.*, **50**, 3104 (1928).

(11) See C. Dufraisse and J. Martel, *Compt. rend.*, **245**, 457 (1957), for a related rearrangement of one of the photooxidation products of 2,4,5-triphenylimidazole.

(12) F. Klingemann and W. F. Laycock, *J. Chem. Soc.*, **59**, 142 (1891).

(13) We thank Prof. F. M. Beringer and Dr. P. S. Forgione for an authentic sample of VI, F. M. Beringer, P. S. Forgione and M. D. Yudis, *Tetrahedron*, **8**, 49 (1960).

(14) G. O. Schenck, *Angew. Chem.*, **60**, 244 (1948), has reported that tribenzoylphenylmethane (VII) is formed in the photooxidation of tetraphenylfuran. In view of earlier confusion of VII with VI [J. Meisenheimer and K. Weibezan, *Ber.*, **54**, 3195 (1921)], it is more probable that Schenck's product, like ours, is the enol benzoate, VI.